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Chapter 9 Modern Chemistry Review Answers

(9-3) 295 Back CHAPTER 9 REVIEW PROBLEMS General Stoichiometry Do not assume that equations without listed coefficients are balanced. 9. Given the chemical equation $\text{Na}_2\text{CO}_3(\text{aq}) + \text{Ca}(\text{OH})_2(\text{s}) \rightarrow 2\text{NaOH}(\text{aq}) + \text{CaCO}_3(\text{s})$, determine to two decimal places the molar masses of all substances involved, and then write them as conversion factors.

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General Chemistry for Engineers explores the key areas of chemistry needed for engineers. This book develops material from the basics to more advanced areas in a systematic fashion. As the material is presented, case studies relevant to engineering are included that demonstrate the strong link between chemistry and the various areas of engineering. Serves as a unique chemistry reference source for professional engineers Provides the chemistry principles required by various engineering disciplines Begins with an 'atoms first' approach, building from the simple to the more complex chemical concepts Includes engineering case studies connecting chemical principles to solving actual engineering problems Links chemistry to contemporary issues related to the interface between chemistry and engineering practices

This book covers the synthesis, reactions, and properties of elements and inorganic compounds for courses in descriptive inorganic chemistry. It is suitable for the one-semester (ACS-recommended) course or as a supplement in general chemistry courses. Ideal for major and non-majors, the book incorporates rich graphs and diagrams to enhance the content and maximize learning. Includes expanded coverage of chemical bonding and enhanced treatment of Buckminster Fullerenes Incorporates new industrial applications matched to key topics in the text

Chemistry, Fourth Edition, by Julia Burdge offers a clear writing style written with the students in mind. Julia uses her experience of teaching hundreds of general chemistry students per year and creates content to offer more in-depth explanation in areas where she knows they have problems. Continuing in the Burdge tradition, the fourth edition maintains an outstanding art program, a consistent problem-solving approach, interesting applications woven throughout the chapters, and a wide range of end-of-chapter problems.

Designed for students in Nebo School District, this text covers the Utah State Core Curriculum for chemistry with few additional topics.

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